SHORT COMMUNICATIONS

Reaction of 2-Vinyloxymethyloxirane and 2-(2-Vinyloxyethoxymethyl)oxirane with Hydrazine

B. F. Kukharev, V. K. Stankevich, N. A. Lobanova, G. R. Klimenko, and E. Kh. Sadykov

Favorskii Irkutsk Institute of Chemistry, Siberian Division, Russian Academy of Sciences, ul. Favorskogo 1, Irkutsk, 664033 Russia e-mail: irk inst chem@irioch.irk.ru

Received December 19, 2005

DOI: 10.1134/S1070428006090284

Reactions of 2-vinyloxymethyl- and 2-(2-vinyloxyethoxymethyl)oxiranes Ia and Ib with ammonia and amines were reported in [1-5]. In the present communication we describe the reaction of compounds Ia and **Ib** with hydrazine. The reaction occurs under mild conditions, and the products (compounds **IIa** and **IIb**) attract interest as monomers and intermediate products in the synthesis of biologically active substances. To obtain compounds **IIa** and **IIb** in a satisfactory yield, the molar ratio hydrazine-epoxy ether I should be no less than 10:1. With a lesser amount of hydrazine, the yield of isolated compounds IIa and IIb strongly decreases, presumably due to formation of thermally unstable products via addition of two epoxide molecules to hydrazine or hydroxy group in **IIa** and **IIb** and their decomposition during vacuum distillation.

$$H_2C$$
 X
 $+$
 H_2NNH_2
 H_2C
 X
 $NHNH_2$
 OH
 IIa, IIb

X = O(a), $OCH_2CH_2O(b)$.

1-Hydrazino-3-(vinyloxy)propan-2-ol (IIa). Compound **Ia**, 10.02 g (0.1 mol), was added dropwise under stirring to 40 g (1 mol) of hydrazine hydrate, maintaining the temperature below 30–35°C, and the mixture was kept for 24 h at room temperature. The mixture was then dried over K_2CO_3 and subjected to vacuum distillation to isolate compound **IIa**. Yield 10.9 g (82.5%), bp 145–147°C (4 mm), $d_4^{20} = 1.0863$,

 $n_D^{20} = 1.4963$. IR spectrum, v, cm⁻¹: 615, 660, 695, 805, 930, 945, 980, 1060, 1100, 1180, 1310, 1445, 1605, 1625, 2855, 2905, 3250–3295. ¹H NMR spectrum, δ, ppm (J, Hz): 2.88 d.d (1H, H_A in NCH_AH_BCH_X, $^2J_{AB} = 12.6$, $^3J_{AX} = 8.0$, $^3J_{BX} = 3.3$), 2.96 d.d (1H, H_B in NCH_AH_BCH_X, $^2J_{AB} = 12.6$, $^3J_{BX} = 8.0$, $^3J_{AX} = 3.3$), 3.34 br.s (4H, OH, NHNH₂), 3.72 d.d (1H, H_A in OCH_AH_BCH_C, $^2J_{AB} = 10.2$, $^3J_{AC} = 5.9$, $^3J_{BC} = 4.8$), 3.77 d.d (1H, H_B in OCH_AH_BCH_C, $^2J_{AB} = 10.2$, $^3J_{BC} = 5.9$, $^3J_{AC} = 4.8$), 4.06 d.d (1H, cis-CH₂=, $^2J = 2.2$, $^3J_{cis} = 6.7$), 4.12 m (1H, CHOH), 4.24 d.d (1H, trans-CH₂=, $^2J = 2.2$, $^3J_{trans} = 14.3$). ¹³C NMR spectrum, δ_C, ppm: 56.49 (NCH₂), 68.51 (OCH₂), 70.19 (CHOH), 87.06 (=CH₂), 151.66 (OCH=). Found, %: C 45.29; H 9.15; N 21.20.

1-Hydrazino-3-[2-(vinyloxy)ethoxy]propan-2-ol (IIb). Compound Ib, 14.42 g (0.1 mol), was added dropwise under stirring to 40 g (1 mol) of hydrazine hydrate, maintaining the temperature below 30°C, and the mixture was kept for 24 h at room temperature. The mixture was then dried over K_2CO_3 and subjected to vacuum distillation to isolate compound IIb. Yield 13.4 g (76%), bp 129–131°C (2 mm), $d_4^{20} = 1.0691$, $n_{\rm D}^{20} = 1.4758$. IR spectrum, v, cm⁻¹: 585, 685, 750, 815, 875, 960, 985, 1060, 1100, 1175, 1225, 1275, 1310, 1345, 1440, 1605, 1620, 2855, 2900, 2910, 3250– 3360. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.22 br.s (3H, NHNH₂), 2.70 d.d (1H, H_A in NCH_AH_BCH_X, $^{2}J_{AB} = 12.9$, $^{3}J_{AX} = 7.1$, $^{3}J_{BX} = 4.1$), 2.81 d.d (1H, H_B in NCH_AH_BCH_X, $^{2}J_{AB} = 12.9$, $^{3}J_{BX} = 7.1$, $^{3}J_{AX} = 4.1$), 3.46 d.d (1H, H_A in OCH_AH_BCH_C, ${}^2J_{AB} = 9.8$, ${}^3J_{AC} =$ 6.7, ${}^{3}J_{BC} = 3.9$), 3.54 d.d (1H, H_B in OCH_AH_BCH_C, ${}^{2}J_{AB} = 9.8$, ${}^{3}J_{BC} = 6.7$, ${}^{3}J_{AC} = 3.9$), 3.70 br.s (1H, OH),

3.71–3.85 m (5H, OCH₂CH₂O, C**H**OH), 4.02 d.d (1H, cis-CH₂=, ${}^{2}J$ = 1.7, ${}^{3}J_{cis}$ = 6.7), 4.19 d.d (1H, trans-CH₂=, ${}^{2}J$ = 1.7, ${}^{3}J_{trans}$ = 14.3), 6.47 d.d (1H, =CH, ${}^{3}J_{cis}$ = 6.7, ${}^{3}J_{trans}$ = 14.3). 13 C NMR spectrum, $\delta_{\rm C}$, ppm: 44.15 (NCH₂), 67.26 (CHOH), 69.80 (=CHOCH₂), 70.98 (=CHOCH₂CH₂O), 73.67 (OCH₂CH), 86.85 (=CH₂), 151.63 (OCH=). Found, %: C 47.87; H 9.56; N 15.73. C₇H₁₆N₂O₃. Calculated, %: C 47.71; H 9.15; N 15.90.

The ¹H and ¹³C NMR spectra were recorded at 26°C on a Bruker DPX-400 spectrometer at 400 and 100 MHz, respectively, using CDCl₃ as solvent and HMDS as internal reference. The IR spectra were obtained on a Specord 75IR spectrophotometer from samples prepared as thin films (neat).

REFERENCES

- 1. Atavin, A.S., Trofimov, B.A., Vyalykh, E.P., and Golovanova, N.I., *Zh. Org. Khim.*, 1969, vol. 5, p. 842.
- 2. Nedolya, N.A., Voronkov, M.G., Zinov'eva, V.P., Dymchenko, V.I., Kositsina, E.I., and Trofimov, B.A., *Zh. Org. Khim.*, 1989, vol. 25, p. 2083.
- 3. Kukharev, B.F., Stankevich, V.K., Klimenko, G.R., Lobanova, N.A., Bel'kova, O.N., Klimenko, E.S., and Timofeeva, S.S., *Zh. Prikl. Khim.*, 1998, vol. 71, p. 1919.
- 4. Kukharev, B.F., Stankevich, V.K., Klimenko, G.R., Lobanova, N.A., Kovalyuk, E.N., and Baranov, A.N., *Zh. Prikl. Khim.*, 2000, vol. 73, p. 161.
- Kukharev, B.F., Stankevich, V.K., Klimenko, G.R., and Lobanova, N.A., *Russ. J. Org. Chem.*, 2000, vol. 36, p. 560.